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Reactions of Thiazolo[3,2-b]-1,2,4-triazolium N-Ylides with Electron-deficient Acetylenes: Novel Benzoyl Migration of Intermediary 1:1-Adducts and Michael Addition to the Acetylenes

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Abstract Reactions of thiazolo[3,2-b]-1,2,4-triazolium N-methylides with electron-deficient acetylenes gave the thiazole ring-opened products as a mixture of the E-and Z-isomers. In contrast, the N-phenacylides reacted with dimethyl acetylenedicarboxylate to give the thiobenzoates together with the thiazole ring-opened products. The mechanism for formation of these compounds is described.

The 1,3-dipolar cycloadditions are very useful reactions to construct new heterocycles. The triazolium N-methylides condensed by a pyridine¹ or a pyrimidine² reacted with dimethyl acetylenedicarboxylate (DMAD) to give the double 1,3-dipolar cycloaddition products. On the other hand, reactions of thiazolium and benzothiazolium N-phenacylides with DMAD gave the thiazole ring-opened products, in which two molecules of DMAD had been incorporated, and the hemithioacetals.³

We envisioned that novel reactivities could be achieved with thiazolo[3,2-b]-1,2,4-triazolium ylides, whose structure is the fused system of a triazole ring and a thiazole ring. The thiazolotriazolium N-ylides 2 were generated in situ from the thiazolotriazolium salts 1 and triethylamine, and allowed to react with equimolar amount of DMAD. The N-phenacylides provided the thiobenzoates 3 and the thiazole ring-opened products 4.4 The structures of these products were determined by spectral data

and X-ray crystallographic data. The formation mechanism for the thiol esters 3 is proposed on the basis of the crossover experiment and the reaction of the α -deuterated phenacyl derivative. The thiobenzoates 3 would be formed via the intramolecular benzoyl migration of the intermediary 1:1-adducts.

The reactions of other N-methylides 2 with 2 equiv. of methyl propiolate (MP) or DMAD afforded the thiazole-ring-opened products 4. The reactions of 2 with MP at room temperature gave better yields than those at 0°C, whereas higher yields were obtained at 0°C in the reactions with DMAD.

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